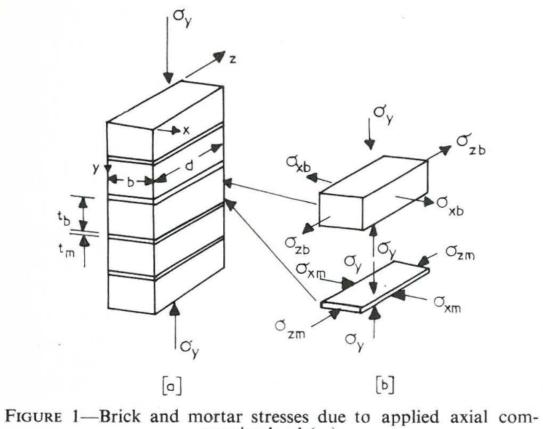
• Triaxial compression...



pressive load (σ_y) .

Extract from: Francis, A.J., Horman, C.B. and Jerrems, L.E., 'The effect of joint thickness and other factors on the compressive strength of brickwork', Proc. of the 2nd Int. Brick Masonry Conf., 1971

- Strength of masonry composite is a function of masonry unit strength, mortar strength, joint thickness, and cross-sectional bond detailing.
- EC6 formula:

$$\mathbf{f}_{ck} = \mathbf{K} \cdot \mathbf{f}_{b,c}^{\alpha} \cdot \mathbf{f}_{m,c}^{\beta}$$

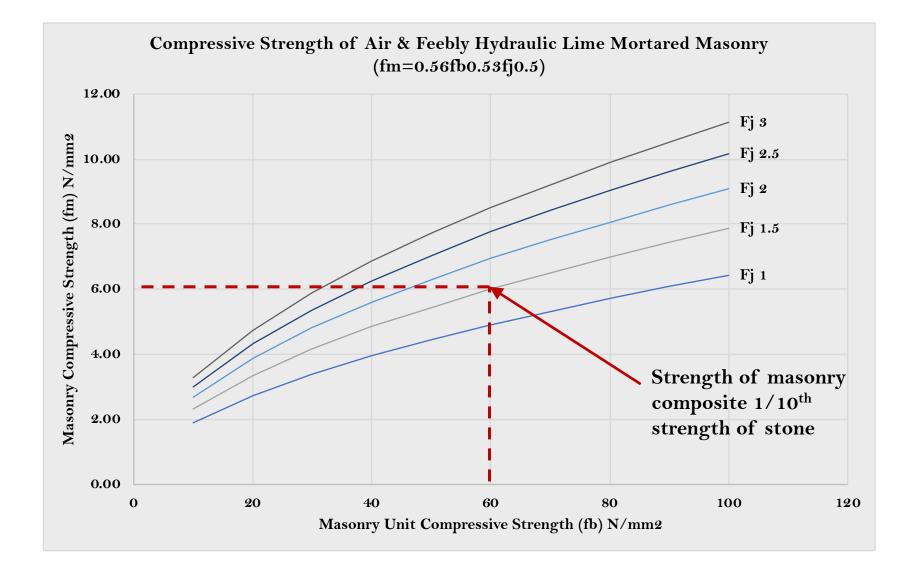
where K would be 0.45 for stone, α is 0.7 and β is 0.3 (UK NA, 2012).

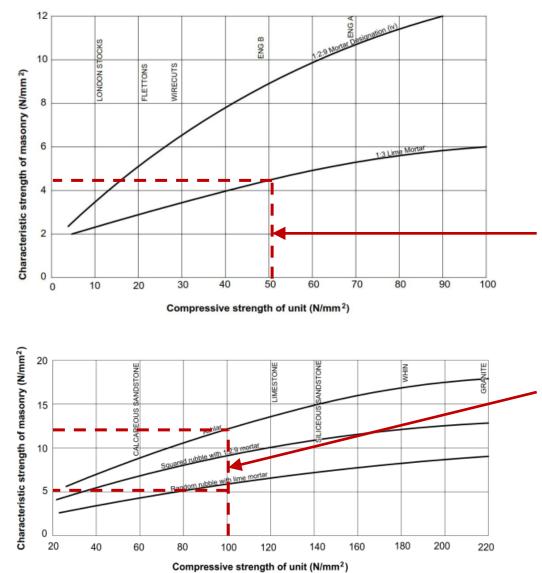
• This is A) a <u>DESIGN</u> code, and B) for <u>CEMENT</u> mortared masonry, and C) for "standard" joint thicknesses

- Strength of masonry composite is a function of masonry unit strength, mortar strength, joint thickness, and cross-sectional bond detailing.
- Costigan et al (2015) corrected formula:

$$f_m = 0.56 \cdot f_b^{\ 0.56} \cdot f_j^{\ 0.5}$$

- For air and feebly hydraulic lime mortared masonry
- For ~12mm "standard" joint thicknesses...
- For 1:3 mix ratios





• <u>Assessment</u> code a bit more instructive than <u>design</u> code

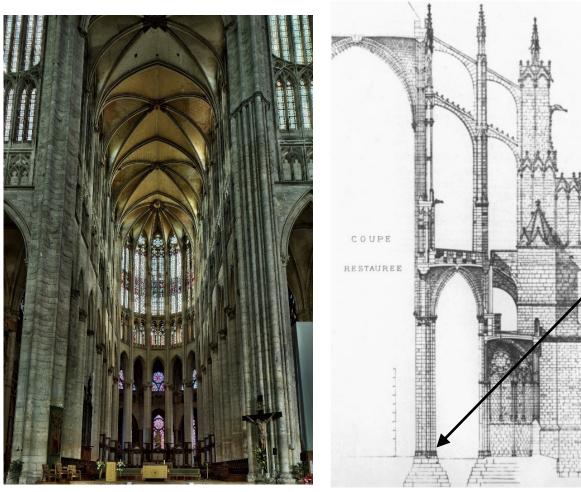
$$\label{eq:eq:theta} \begin{split} \theta &\leq 0.6 \, \bullet \, f_k / \ \left(\gamma_m {=} 1.5 \right) \\ \text{(CD 376 cl. 4.7)} \end{split}$$

Strength of masonry composite about $1/10^{\text{th}}$ strength of the brick for median values

Random rubble ca. 1/20th strength of the stone, significantly less strong than ashlar for the same stone type* (*mortar for ashlar work not recorded)

Extracts from: CS 454 – Assessment of Highway Bridges and Structures (DMRB, 2019)

• Gothic cathedrals...



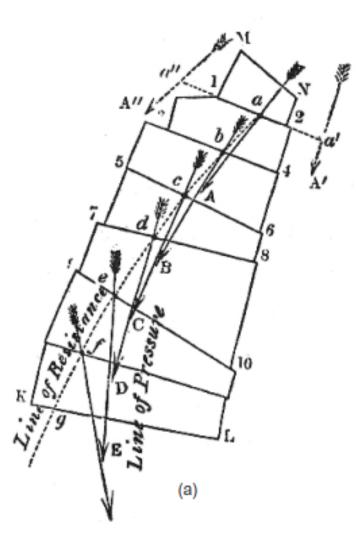
- Historically stresses were consciously limited to 1/10th the crushing strength of the stone
- Beauvais: compressive stress in base of crossing pier around **1.3 N/mm²**
- (see Jacques Heyman The Stone Skeleton)

• High load, low stress

'Safe Theorem':

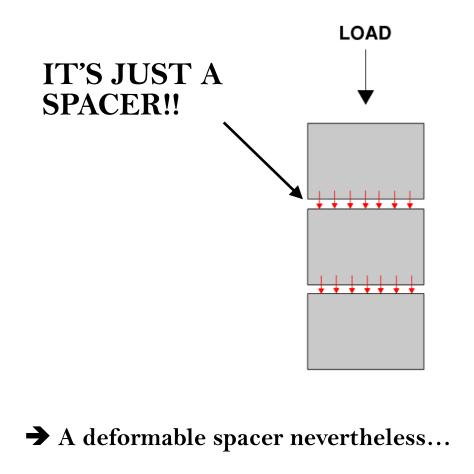
- Unlimited compressive strength
- Sliding does not occur
- Zero tensile strength

[The Safe Theorem – J. Heyman, 1995]

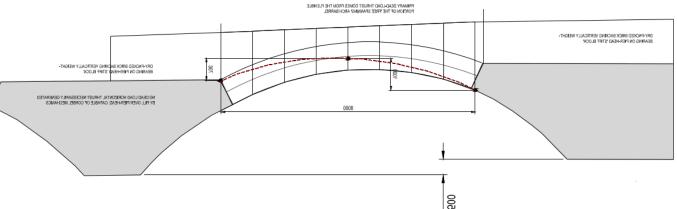


*The Stone Skeleton (Heyman, 1995) has got to be on any self-respecting conservation engineer's bookshelf...

• The role of the mortar joint...







Deformability:

- Described by Young's Modulus
- Varies significantly according to joint thickness and binder type
- An expression for approximating the elastic modulus of masonry is given in EN 1996-1-1:2005(+A1: 2012):

$$\mathbf{E} = \mathbf{K}_{\mathbf{E}} \cdot \mathbf{f}_{\mathbf{k}}$$
 where K_E would be 1,000 (UK NA).

- Formula relates compressive strength of the masonry composite with stiffness by a constant
- Intended application of the code is **CEMENT**-mortared masonry

• Grossly over-estimates the modulus of lime-mortared masonry

Deformability:

• Costigan et al. (2015) corrected formula (non/feebly hyd'c):

 $E = 85 \cdot f_m$

• Ditto for NHL 3.5/5 eminently hydraulic:

 $E = 130 \cdot f_m$

• (EC6 for cement mortared masonry):

 $E = 1,000 \cdot f_m$

MASONRY BUILT WITH TRADITIONAL LIME MORTAR IS <u>AN ORDER OF MAGNITUDE</u> <u>MORE DEFORMABLE</u> THAN CEMENT MORTARED MASONRY

• Oft-neglected mechanical role...

- Real buildings creak, groan and wobble around
- Heavy structures on rubbish foundations
- Lego bricks, jointed in plasticine, bearing on play-dough



• Deformable cushion

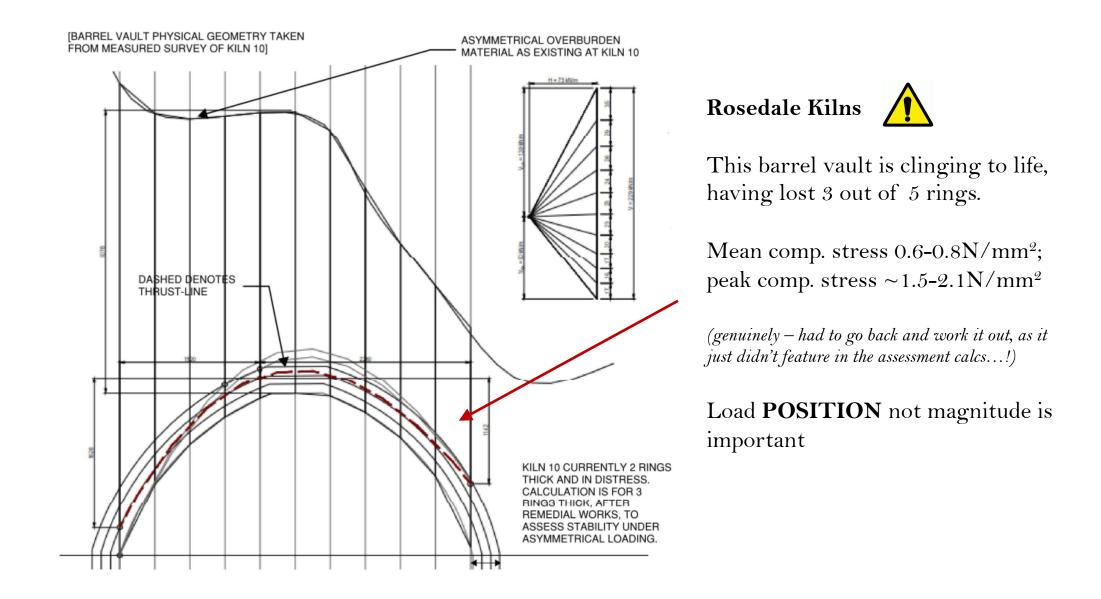
This one's dangerous...

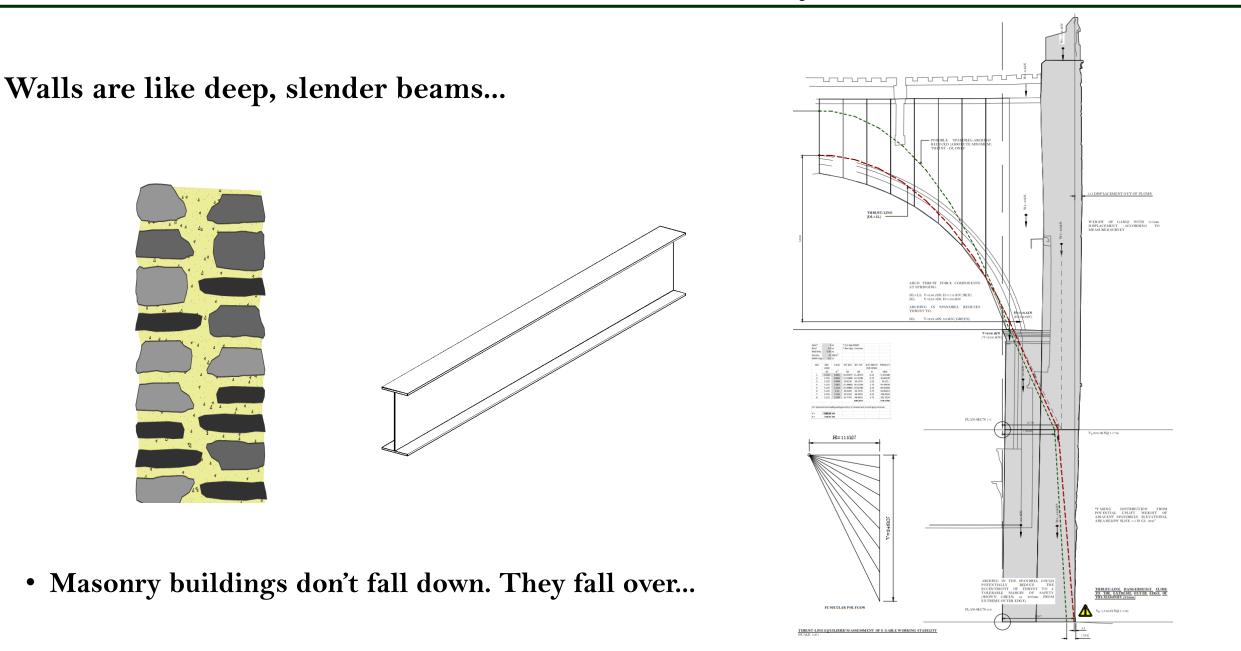






See the 3D model of the kiln here: https://sketchfab.com/3d-models/rosedale-east-stone-kilns-frontage-7c28dc01d8e84bdc9dbfac0a7dca9ff2





• Daft question #1...



Lime mortar is simply a low-stress deformable packer between masonry units.

• Daft question #2...



Carbonation:

- 1. Carbon dioxide diffuses into pore space of mortar;
- 2. CO_2 dissolves in pore water to form carbonic acid H_2CO_3 ;
- 3. Free lime $CaOH_2$ dissolves in pore water (reduced ph);
- 4. Carbonic acid carbonate ions react with calcium ions from the lime, to form calcium carbonate $CaCO_3$.

Word to the wise: moisture is essential to carbonation process. BUT!

- Carbonation 'impossible' below 20% RH...
- Carbonation impossible at 100% RH...
- Optimum RH ~60%
- Lab environment?
- Meanwhile in Scotland

AIR WITHIN PORES OF DAMP MATERIALS PRACTICALLY 100% RH...





Hydraulic mortars:

- 1. Hydraulic limestone source or pozzolanic mortar;
- 2. Calcium silicates and aluminates react with water molecules to form calcium silicate & aluminate hydrates
- 3. i.e. "Hydration". 2CaOSiO₂ (Belite), 3CaOSiO₂ (Alite), 3CaOAl₂O₃ (Tricalcium aluminate)...
- 4. This chemical set gives initial hardening and early strength;
- 5. Thereafter, carbonation takes place

Or so the story goes...



Gauged lime mortars:

- To make them strong
- To make them harden in damp/wet
- To resist frost

To get the best of both worlds?

- So we use a mortar that needs to stay wet to hydrate, but needs to be dry to carbonate...
- 1. The early set lulls us into a sense of false security
- 2. We think it's 'went off', it's exposed to the elements, probably wet to begin with...
- 3. But the mortar's still overwhelmingly air lime and this means no carbonation
- 4. The more hydraulic the mix, the slower the carbonation rate...
- 5. The wetter the substrate is, and the more water we add to it by 'aftercare', the longer the delay in carbonation will be...

